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Section 2.2 mixtures.

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STUDY. PLAY. Mixture. is a physical blend of two or more components.

Heterogeneous mixture. is a mixture that is not uniform in composition. Solution. is a homogeneous mixture. Phase. is used to describe any part of a sample with uniform composition and properties. Filtration.

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Answers of oxygen to hydrogen in  $H_2O_2$  is two times the mass ratio of oxygen to hydrogen in  $H_2O$ . This follows the law of multiple proportions.

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54–58) This section  
discusses chemical  
properties and  
describes clues that  
may show that a  
chemical change has  
taken place. Reading  
Strategy (page 54)  
Relating Text and  
Visuals As you read,  
complete the table by

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finding examples of the clues for recognizing chemical changes in Figures 19 and 20.

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479 1. Explain Use the  
properties of seawater

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to describe the characteristics of mixtures. Answers will vary but might include that seawater is a heterogeneous mixture with dirt

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donkey together” constitutes a separate prohibition that refers to literal “plowing”. The prohibition on literal “plowing” immediately follows the prohibition on kilayim of plants because it is also about fieldwork, not because it is also about kilayim.

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practical activity in the 'States of matter and mixtures' section:  
Investigate the composition of inks using simple distillation and paper chromatography.  
Sample question 1 ...

### **Sample exam questions - States of matter and mixtures - BBC**

Section 2.2 Physical Properties (pages 45-51) This section

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discusses physical properties and physical changes. It also explains how physical properties can be used to identify materials, select materials, and separate mixtures.

Reading Strategy (page 45) Building Vocabulary As you read, write a definition for each term in the table below.

## **Chapter 2 Properties of Matter Section**

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## **2.2 Physical Properties**

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43 Elements,  
Compounds, and  
Mixtures SECTION 3

Name Class Date

Mixtures continued Do  
Mixtures Have Fixed  
Ratios? A compound is  
made of elements in a  
fixed, or specified,  
ratio. For example,  
water is always two  
parts hydrogen and  
one part oxygen. A

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mixture, however, does not have a fixed ratio of components.

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