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7.9d We consider here a chemical model for the acidification of fog droplets (Hoffmann and Jacob 1984). Fog droplets are assumed to form around cloud condensation nuclei (CCN) such as NaCl. At time  $t = 0$  droplets are assumed to condense suddenly in an atmospheric "closed box" containing CCN and the

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trace gases  $\text{CO}_2$ ,  $\text{SO}_2$ ,  $\text{NH}_3$ ,  $\text{HNO}_3$ ,  $\text{H}_2\text{O}_2$ , and  $\text{O}_3$ .

## **Appendix Oxygen And Hydrogen Chemistry - Atmospheric Chemistry**

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Daniel J. Jacob is the Gordon McKay Professor of Atmospheric Chemistry and Environmental Engineering at Harvard University. He has taught the undergraduate atmospheric chemistry course at Harvard since 1992. He has published over 100 research papers in atmospheric chemistry journals.

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